

New Problems Chapter 20

- 20.1-7. Diffusion through a solid using Fick's second law.** Determine the carburizing time necessary to achieve a carbon concentration of 0.5% at a position 2.5 mm into an iron-carbon alloy that initially contains 0.10% C. The surface concentration is to be maintained at 1.5% C, and the treatment is to be conducted at 1000°C with a $D_{AB} = 6.2 \times 10^{-11} \frac{m^2}{s}$. Use Fick's second law to solve the equation:

$$\frac{C_x - C_0}{C_s - C_0} = 1 - \operatorname{erf}\left(\frac{x}{2\sqrt{D_{AB}t}}\right)$$

- 20.1-8. Diffusion in a solid using Table 20.1.1.** Walmart is doing a study on the shelf life of Italian sausage (cylindrical) in its super market. Their sausages are 2in in diameter and 6in long and the case is at 10°C. Initially the sausage has a liquid moisture content of 0.001M, but sitting in the meat case it is exposed air with a concentration of water $C_w = 4.30 \frac{\text{mole}}{m^3}$. The equilibrium constant between the air and the sausage is $K = 1.5$ and assume $D_{AB} = 1 \times 10^{-5} \frac{cm^2}{s}$ and $k = \infty$. How long (hours) will it take for the center of the sausage to reach 2.12 times its initial concentration.

- 20.2-2. Unsteady-State Diffusion and Reaction.** Solute A is diffusing at unsteady state into a semi-infinite medium of pure B and undergoes a 1st order reaction with B. Solute A is dilute. How long will it take for the concentration $C_A = .2M$ at $x = 0.4mm$? Physical Property data are $D_{AB} = 1 \times 10^{-9} \frac{m^2}{s}$, $k' = 1 \times 10^{-4} \frac{1}{s}$, $C_{A0} = 1M$.

- 20.3-4. Unsteady-State Diffusion and Reaction (COMSOL).** A flat slab of wood, with a height of 100 mm and a width of 50 mm, contains $3 \frac{\text{mole}}{m^3}$ moisture is being dried from both sides (ends are insulated). The equilibrium moisture content at the surface of the wood due to the drying air is held at a $0.1 \frac{\text{mole}}{m^3}$ moisture content. If $D_{AB} = 1.033 \times 10^{-9} \frac{m^2}{s}$ how long will it take the center to reach a $1 \frac{\text{mole}}{m^3}$ moisture content.